Name	KEY	98	
Signature			

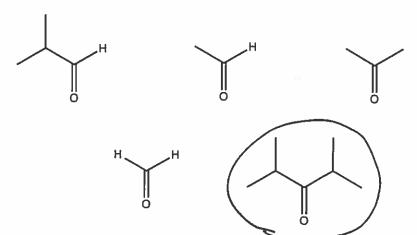
You are required to answer all question sets. Please note that some of the question sets offer you a choice of questions - do only the number of questions asked for. Please write legibly and draw clearly. Points will be deducted for illegible and unclear answers. The point total for this exam is 100 and the value of each question set is shown in parenthesis beside that set. <u>READ ALL QUESTIONS CAREFULLY AND APPORTION YOUR TIME ACCORDINGLY.</u>

PART ONE: STRUCTURE, REACTIVITY, AND MECHANISM

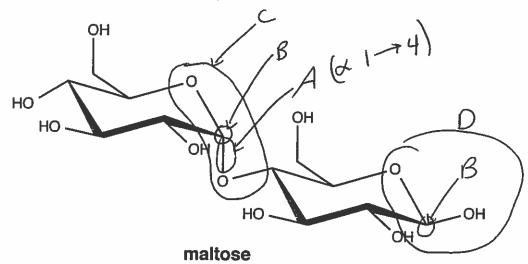
1. Which reactions(s) (A-E) below would NOT occur as written? Circle all that apply.

(6)

2. Of the five structures below, circle the *one* structure that you would expect to have the lowest % hydrate at equilibrium after treatment with catalytic OH in water.



3. Consider the D-glucose disaccharide below, maltose.



On the structure of maltose (above) circle and label each of the following:

(5)

(4)

A. The glycosidic bond. Which of the following correctly describes the bond? Check your answer.

- B. An anomeric carbon atom.
- C. An acetal functionality.
- D. A hemiacetal functionality.

4. A trans-esterification reaction follows essentially the same mechanism as a Fischer esterification, only with a different leaving group. Use what you know about Fischer esterification, and the behavior of acid-catalyzed mechanisms in general, to draw the stepwise mechanism for the following transesterification. You must account for the formation of all products to receive full credit. (14)

<u>PART TWO: PREDICT THE PRODUCT</u> - For *eight* (8) of the following questions 5-15 predict the *major* organic product(s) showing stereochemistry where appropriate. (32)

Check the corresponding boxes for the EIGHT answers that you want graded, otherwise **5-12 will be graded automatically**.

5.
$$\begin{array}{c|c} Ph \\ Ph \\ \hline Ph \\ \hline \end{array}$$

16. Fill in the missing reagents and/or products in the synthetic scheme outlined below. Place your answers in the table provided on the next

MISSING REAGENT/PRODUCT		Soch	0=\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	0=\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
		7	Y	
MISSING REAGENT/PRODUCT	1. Co ₂ 2. H ₃ 0+	Ch A Leas	H MnOy	1. L: AIHY 2. H20
	ш	Ľ.	9	Ι
MISSING REAGENT/PRODUCT	H H	PCC	# 1	X = X Or BY
	4	ω	U	a

OR

One possible answer for each Shown (many other answers would work as well)

$$\begin{array}{c}
 & 2Li \\
\hline
 & 2Li
\end{array}$$

$$\begin{array}{c}
 & Li \\
\hline
 & 2. H_3 o^t
\end{array}$$

$$\begin{array}{c}
 & O \\
\hline
 & Piece \\
\end{array}$$

$$\frac{CH_2 = PPh_3}{2. H_2O_2/OH} - OH$$

$$Piece Z$$